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Secretions of the Stomach Part-

Ingested food and stomach secretions, mixed together, form a semi fluid material called **chyme**. The stomach functions primarily as storage and mixing chamber for the chyme. Although some digestion and absorption occur in the stomach, they are not its major functions. Stomach secretions include mucus, hydrochloric acid, gastrin, histamine, intrinsic factor, and pepsinogen. Pepsinogen is the inactive form of the protein-digesting enzyme pepsin.

The surface mucous cells and mucous neck cells secrete viscous and alkaline **mucus** that covers the surface of the epithelial cells and forms a layer 1-1.5 mm thick. The thick layer of mucus lubricates and protects the epithelial cells of the stomach wall from the damaging effect of the acidic chyme and pepsin. Irritation of the stomach mucosa results in stimulation of the secretion of a greater volume of mucus. Parietal cells in the gastric glands of the pyloric region secrete intrinsic factor and a concentrated solution of hydrochloric acid.

Intrinsic factor is a glycoprotein that binds with vitamin B12 and makes the vitamin more readily absorbed in the ileum. Vitamin B12 is important in deoxyribonucleic acid (DNA) synthesis.

Hydrochloric acid produces the low pH of the stomach, which is normally between 1 and 3. Although the hydrochloric acid

Secreted into the stomach has a minor digestive effect on ingested food, one of its main functions is to kill bacteria that are ingested with essentially everything humans put into their mouths. Some pathogenic bacteria may avoid digestion in the stomach, however, because they have an outer coat that resists stomach acids. The low pH of the stomach also stops carbohydrate digestion by inactivating salivary amylase. Stomach acid also denatures many proteins so that proteolytic enzymes can reach internal peptide bonds, and it provides the proper pH environment for the function of pepsin.

Hydrogen ions are derived from carbon dioxide and water, which enter the parietal cell from its serosal surface, which is the side opposite the lumen of the gastric pit. Once inside the cell,

carbonic anhydrase catalyzes the reaction between carbon dioxide and water to form carbonic acid. Some of the carbonic acid molecules then dissociate to form hydrogen ions and bicarbonate ions. The hydrogen ions are actively transported across the mucosal surface of the parietal cell into the lumen of the stomach; some potassium ions are moved into the cell in exchange for the hydrogen ions. Although hydrogen ions are actively transported against a steep concentration gradient, chloride ions diffuse with the hydrogen ions from the cell through the plasma membrane. Diffusion of chloride ions with the positively charged hydrogen ions reduces the amount of energy needed to transport the hydrogen ions against both a concentration gradient and an electrical gradient. Bicarbonate ions move down their concentration gradient from the parietal cell into the extracellular fluid. During this process, bicarbonate ions are exchanged for chloride ions through an anion exchange molecule, which is located in the plasma membrane, and the chloride ions subsequently move into the cell.

- Carbon dioxide (CO₂) diffuses into the cell.
- CO₂ is combined with water (H₂O) in an enzymatic reaction that is catalyzed by carbonic anhydrase (CA) to form carbonic acid (H₂CO₂).
- Carbonic acid dissociates into a bicarbonate ion (HCO₃) and a hydrogen ion (H⁺).
- HCO₃⁻ is transported back into the bloodstream. An anion exchange molecule in the plasma membrane exchanges HCO₃⁻ for a chloride ion (Cl⁻) (counter transport).
- The hydrogen ion (H⁺) is actively transported into the duct of the gastric gland.
- Chloride ions (CI⁻) diffuse with the charged hydrogen ions.
- Some potassium ions (K⁺) are counter transported into the cell in exchange for the hydrogen ions.



Hydrochloric Acid Production by Parietal Cells in the Gastric Glands of the Stomach